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3-1 Calculate the specific volume of an air-vapor mixture in cubic meters per kilogram of dry air when the following conditions prevail: $t = 30\text{ C}$, $W = 0.015\text{ kg/kg}$, and $p_t = 90\text{ kPa}$. Solution: Equation 3-4. $t_s = t_a = t_p = T = 30\text{ C} + 273 = 303\text{ K}$ $R_a = 287\text{ J/kg.K}$ $p_t = 90\text{ kPa} = 90,000\text{ Pa}$ Equation 3-2 $t_s = p_t \cdot W = 90 \cdot 0.015 = 1.35$ $t_s = 1.35 - 0.15p_s$

or [CLICK HERE](#)

Solution: $q = mc_p(t_2 - t_1)$ $m = 2.5\text{ kg/s}$ $c_p = 1.0\text{ kJ/kg.K}$ $t_2 = 30\text{ C}$ $t_1 = -10\text{ C}$ CHAPTER 2 - THERMAL PRINCIPLES Page 2 of 5 Then, $q = (2.5)(1.0)(30 + 10)$ $q = 100\text{ kw}$ --- Ans. 2-3.

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