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ATOM (pages 127–132) Chapter 5
Electrons in Atoms 43 SECTION 5.1
MODELS OF THE ATOM (pages
127–132) This section summarizes the
development of atomic theory.

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Chapter 5 Electrons in Atoms 43 Answer

SECTION 5.1 MODELS OF THE ATOM

(pages 127–132) This section summarizes the development of atomic theory. It also explains the significance of quantized energies of electrons as they relate to the quantum mechanical model of the atom.

The Development of Atomic Models

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(pages 127–128) 1. Atoms Answer
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SECTION 5.1 MODELS OF THE ATOM

(pages 127–132)

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SECTION 5.2 ELECTRON

ARRANGEMENT IN ATOMS (pages

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5.1 Revising the Atomic Model >

□ According to the Pauli exclusion principle, an atomic orbital may describe at most two electrons. □ To occupy the same orbital, two electrons must have opposite spins; that is, the electron spins must be paired.

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5.2 Electron Arrangement in Atoms >

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