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Introduction to Organic Chemistry and Biochemistry. Part I - Organic Chemistry. Hydrocarbons are molecules that contain only hydrogen and carbon atoms Each Carbon atom forms 4 bonds and each hydrogen forms 1 bond. Hydrocarbons include

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(among other things) alkanes - all single bonds between carbons
alkenes - one or more double bonds between carbons
alkynes - one or more triple bonds between carbons
aromatics - contain benzene ring or similar structures.

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Phương Trần - Academia.edu Carbohydrates are the most abundant organic compounds in the plant world. They act as storehouses of chemical energy (glucose, starch, glycogen); are components of supportive structures in plants (cellulose), crustacean shells (chitin), and

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Introduction to Organic and Biochemistry (William H. Brown and Lawrence S. Brown) 8th Edition. by Frederick A. Bettelheim (Author), William H. Brown (Author), Mary K. Campbell (Author), Shawn O. Farrell (Author), Omar Torres (Author) & 2 more. 3.9 out of 5 stars 10 ratings. ISBN-13: 978-1133109761.

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Frederick A. Bettelheim, William H. Brown, Mary K. Campbell, Shawn O. Farrell. Cengage Learning, Jan 5, 2009 - Science - 864 pages. 4 Reviews. This bestselling text continues to lead the way with a strong focus on current issues; pedagogically rich framework; wide variety of medical and biological applications; visually dynamic art program; and

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exceptionally strong and varied end-of-chapter problems.

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Introduction to General, Organic, and Biochemistry, Tenth Edition has clear & logical explanations of chemical concepts and great depth of coverage as well as a clear, consistent writing style which provides great readability. An emphasis on Real-World aspects of chemistry makes the reader comfortable in seeing how the chemistry will apply to their career.

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See an explanation and solution for Chapter 13,
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by ...

About the Book The Basics of General, Organic, and Biological Chemistry by David W. Ball, John W. Hill, and Rhonda J. Scott is for the one-semester General, Organic and Biological Chemistry course. The authors designed this textbook from the ground up to meet the needs of a one-semester course.

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