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Synthesis: □ Two or more substances combine to form a more complex substance. □  $A + B = AB$  □  $Fe ( s ) + S(s) = FeS(s)$  □  $H_2 O ( l ) + SO_3 ( g ) = H_2 SO_4 ( aq )$  2.

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Formula  $\square$  represents the \_\_\_\_\_ number of  
\_\_\_\_\_ of each

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doz grains of rice = \_\_\_\_\_g (Use this fact  
as a conversion

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R. Lab results: 1 doz grains of rice = 0.3 g  
(Use this fact as a conversion factor.)?

grains of rice = 1.94 g Avogadro's

Number - the mole = the number  $6.022 \times 10^{23}$   
Molar Mass the mass of one mole  
of any pure substance



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