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Equilibria and Buffer

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Thermofluids 1 Chapter 5

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csilk. Terms in this set

(47) characteristics of the water molecule. 1) triatomic
2) each O-H bond is highly

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polar b/c of oxygen's high
electronegativity, water is
a whole is a polar molecule

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Chapter 17 Water Aqueous

Systems€1) water molecules

in constant motion, collide

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with salt crystals 2) water molecules attracted to the anions and cations in the salt due to the water's partial +/- charges - H attracts Chapter 17 Water Aqueous Systems Practice Problems Answers chapter 17

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in constant motion, collide
with salt crystals 2)water
molecules attracted to the
anions and cations in the

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Water and its

Properties OBJECTIVES: •

Explain the high surface
tension and low vapor
pressure of water in terms
of the structure of the

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Practice molecule and hydrogen bonding.

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Review. • Water is a simple tri-atomic molecule, H_2O . • Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values ...

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