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Answer. Molar mass = 151.1 g/mol Potassium: $(39.10 / 101.1) \times 100 = 38.67\%$ Nitrogen: $(14.01 / 101.1) \times 100 = 13.86\%$ Oxygen: $(48.00 / 101.1) \times 100 = 12.58\%$. Molar mass = 164.1 g/mol Potassium: $(39.10 / 101.1) \times 100 = 38.67\%$ Nitrogen: $(14.01 / 101.1) \times 100 = 18.96\%$ Oxygen: $(48.00 / 101.1) \times 100 = 47.48\%$.

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